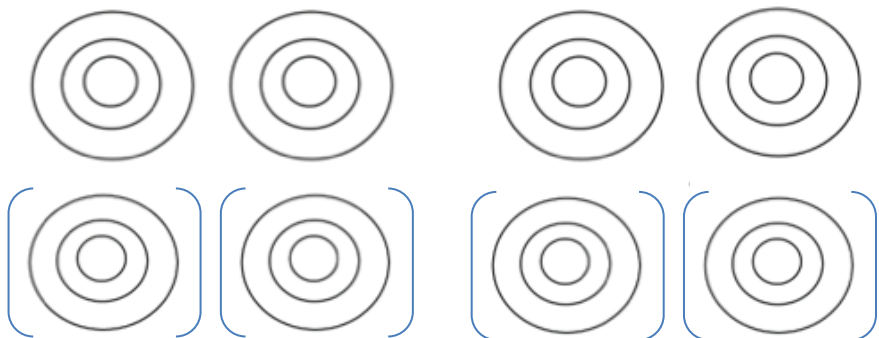


## Ionic Bonding

- 1) Fill in the electron configuration for the atoms.
- 2) Draw an arrow to show how the electron moves.
- 3) Draw the new ions and add the charge to the square brackets.

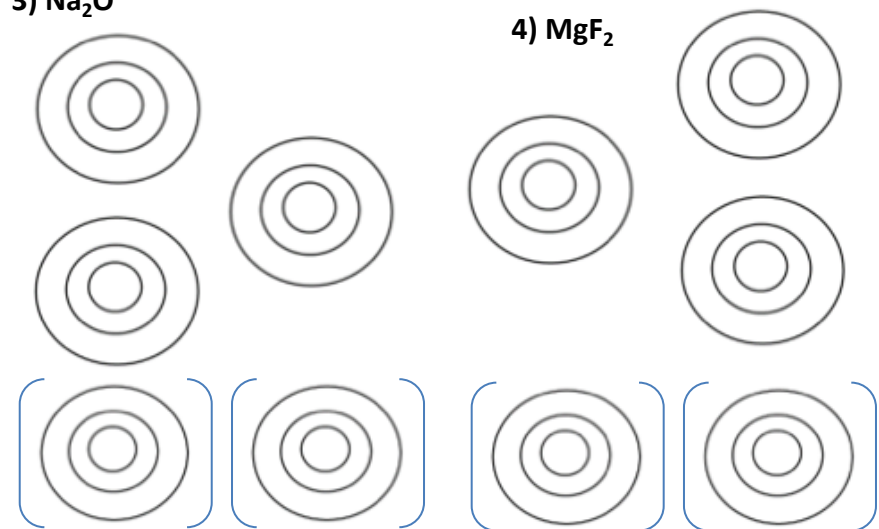
1) MgO

2) LiCl



3) Na<sub>2</sub>O

4) MgF<sub>2</sub>



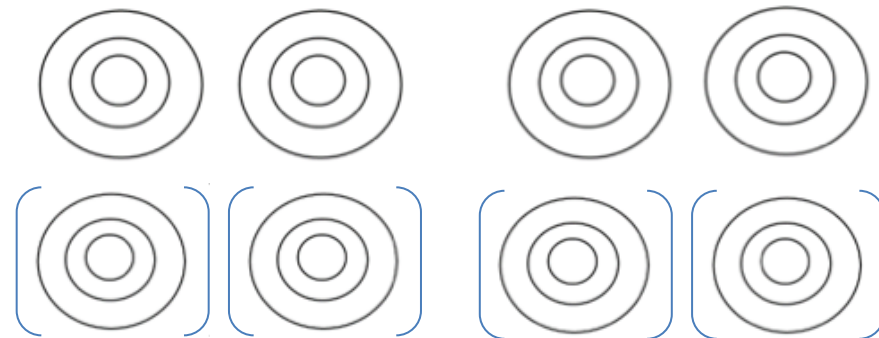
5) Form the ionic bonds for: AlCl<sub>3</sub> NaF, Al<sub>2</sub>O and Li<sub>2</sub>O.

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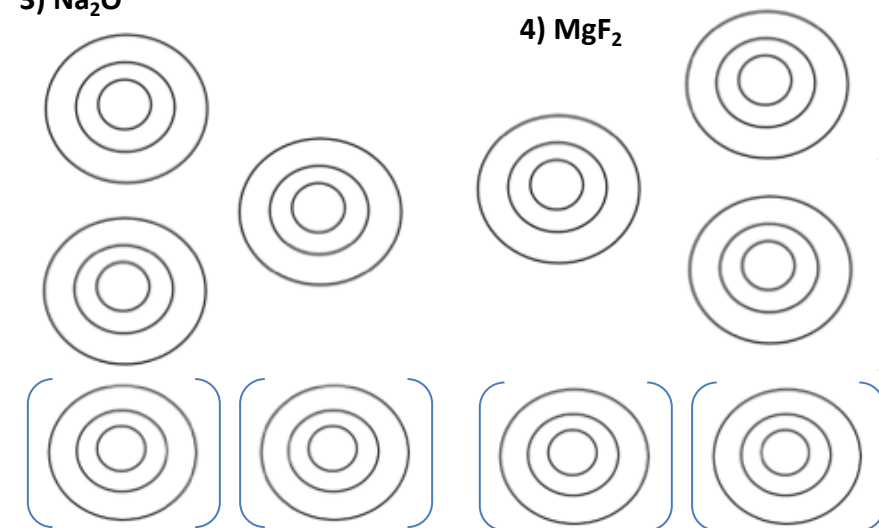
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5) Form the ionic bonds for: AlCl<sub>3</sub> NaF, Al<sub>2</sub>O and Li<sub>2</sub>O.